

ANSWERS: Crystal ball questions on Lewis structures practice

N_2 	H_2O_2 	HOCl 	F_2 or Cl_2 or Br_2 or I_2 or replace the Cl atoms with F, Br or I atoms
BF_3 or BCl_3 	NH_3 	C_2H_4 	CF_4
CCl_4 	BeCl_2 	SiH_4 	Br_2
NH_2OH 	C_2H_4 	HNO_2 	P_4
C_2H_2 	SCl_2 	O_3 	SO_3
N_2H_4 	CO 	expanded octet: the S central atom has more than 8 electrons, this should not be a level 2 NCEA Exam question (but SO_2 appeared in NCEA 2010 Level 2 Exam)	